# Biology Year 9

# Organic Molecules

Name: \_\_\_\_

Form: \_\_\_\_\_

## Introduction

- 1. What is the basic element of all organic material?
- 3. Draw a diagram of the organic molecule

4. List the four classes of organic macromolecules.

5. A polymers is a \_\_\_\_\_ molecule consisting of \_\_\_\_\_ sub units called \_\_\_\_\_ bonded together



- The basic structure of a carbohydrate is a \_\_\_\_\_\_ molecule combined with a \_\_\_\_\_\_ molecule to form the molecular formula of \_\_\_\_\_\_. The two common types of carbohydrates are \_\_\_\_\_\_ and \_\_\_\_\_\_. The function of carbohydrates are to store \_\_\_\_\_\_ energy for cellular use and give \_\_\_\_\_\_ in plants.
- Monosaccharides (simple sugar) have the formula \_\_\_\_\_\_where n can be \_\_\_\_\_\_.
  The most common and important monosaccharide is \_\_\_\_\_\_\_, which is a six-carbon, so has the formula C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>. Its structure is:



The other two monosaccharide's are \_\_\_\_\_ and \_

#### 3. Disaccharides (double sugars) are formed when \_\_\_\_\_

joined together by a <u>glycosidic bond</u>. The reaction involves the formation of a molecule of water

There are three common disaccharides:

- \_\_\_\_\_\_ (or malt sugar) is\_\_\_\_\_\_. It is formed on digestion of starch by amylase, because this enzyme breaks starch down into two-glucose units. Brewing beer starts with <u>malt</u>, which is a maltose solution made from germinated barley.
- \_\_\_\_\_\_ (or cane sugar) is \_\_\_\_\_\_. It is common in plants because it is less reactive than glucose, and it is their main transport sugar. It is the common table sugar that you put in your tea.
- \_\_\_\_\_\_ (or milk sugar) is \_\_\_\_\_\_. It is found only in mammalian milk, and is the main source of energy for infant mammals.

#### 4. Polysaccharides

Polysaccharides are long chains of \_\_\_\_\_\_.

There are three important polysaccharides:

- \_\_\_\_\_\_\_ is the plant storage polysaccharide. It is insoluble and forms starch granules inside many plant cells. Being insoluble means starch does not change the water potential of cells, so does not cause the cells to take up water by osmosis.
- \_\_\_\_\_is only found in plants, where it is the main component of cell walls.
- \_\_\_\_\_\_is the storage molecule of sugar that is stored in the liver



НÓ b1-4 glyclosidic bonds in cellulose

	Lipids			
	Lipids			
	complex simpl	e		
	Triglycerides Phospholipids Waxes Steroids	Terpenes		
1.	The basic characteristics of a lipid contains	oxygen's than a		
	carbohydrate			
2.	Give examples of the various lipids.			
3.	The function of lipids are long term, ,,,,	structure, and		
4.	Lipids are made up of and			
 po	are long molecules with a polar, lar, "tail".	end and a non-		
Hydrophilic means the molecule is wl		_ while hydrophobic		
me	means the molecule is			

• If †	here are no C=C double bonds in the hydro	carbon chain, then it is a		
	. These fatty acids form	chains, and have a		
	melting point. Most of these are	fats		
$\sim$	$ \frown \frown$	<u> </u>		
• Ift	If there are C=C double bonds in the hydrocarbon chain, then it is an $\_$			
	These fatty acids f	form chains, and		
have a	melting point. Fatty acids with r	more than one double bond		
are called	Most d	of these are		
	fats			

### Proteins



\_\_\_\_\_\_ are basic polymer of a protein. They are made of \_\_\_\_\_\_\_ and \_\_\_\_\_\_.
 These polymers are held together with a \_\_\_\_\_\_ bond. This is why proteins are called \_\_\_\_\_\_.

#### **Basic Function of Proteins**

structure	e.g. collagen (bone, cartilage, tendon), keratin (hair), actin (muscle)
enzymes	e.g. amylase, pepsin, catalase, etc (>10,000 others)
transport	e.g. haemoglobin (oxygen), transferrin (iron)
pumps	e.g. Na⁺K⁺ pump in cell membranes
motors	e.g. myosin (muscle), kinesin (cilia)
hormones	e.g. insulin, glucagon
receptors	e.g. rhodopsin (light receptor in retina)
antibodies	e.g. immunoglobulins
storage	e.g. albumins in eggs and blood, caesin in milk
blood clotting	e.g. thrombin, fibrin
lubrication	e.g. glycoproteins in synovial fluid
toxins	e.g. diphtheria toxin

antifreeze e.g. glycoproteins in arctic flea and many more!

How many amino acids are there? (they are determined by the R group)

EXAMPLE OF AN AMINO ACID SEQUENCE

 $^{+}NH_{3}$ -Gly — Pro — His — Leu — Tyr — Ser — Trp — Asp — Lys — Cys-COO<sup>-</sup>

#### Protein Structure

Polypeptides are just a string of amino acids, but they fold up to form the complex and well-defined three-dimensional structure of working proteins.:

1. Label the four shapes of protein?



#### determine protein conformation?